

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

N,N'-Bis(2-methoxyphenyl)anthracene-1,8-dicarboxamide

En Gao, Rui He, Huan-Xia Zhang and Da-Bin Qin*

College of Chemistry and Chemical Engineering, China West Normal University, Nanchong 637002, People's Republic of China
Correspondence e-mail: qindabincwnu@yahoo.com.cn

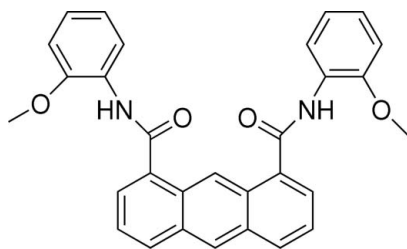
Received 30 July 2007; accepted 8 August 2007

Key indicators: single-crystal X-ray study; $T = 153$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.106; data-to-parameter ratio = 13.2.

In the title compound, $\text{C}_{30}\text{H}_{24}\text{N}_2\text{O}_4$, the anthracene system forms dihedral angles of 59.26 (5) and 82.83 (5)° with the benzene rings of the methoxyphenyl groups. Another feature of interest is the dihedral angle between the methoxyphenyl benzene rings, which is 42.62 (5)°. In the molecule, there are four very weak intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds and the crystal packing is stabilized by weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For synthesis, see: Rogers & Averill (1986); Gunnlaugsson *et al.* (2005). For properties of anthracene-based molecules, see: Chen & Chen (2004); Gunnlaugsson *et al.* (2003).



Experimental

Crystal data

$\text{C}_{30}\text{H}_{24}\text{N}_2\text{O}_4$
 $M_r = 476.51$
Monoclinic, $P2_1/c$
 $a = 13.131$ (3) Å
 $b = 11.864$ (2) Å

$c = 14.911$ (3) Å
 $\beta = 91.13$ (3)°
 $V = 2322.4$ (8) Å³
 $Z = 4$
Mo $K\alpha$ radiation

$\mu = 0.09$ mm⁻¹
 $T = 153$ (2) K

$0.30 \times 0.28 \times 0.28$ mm

Data collection

Rigaku R-Axis RAPID diffractometer
Absorption correction: none
18982 measured reflections

4318 independent reflections
3977 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.016$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.107$
 $S = 1.03$
4318 reflections

328 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.45$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C14—H14 ⁱ ···O1	0.93	2.51	3.0581 (18)	118
C14—H14 ⁱ ···O3	0.93	2.55	3.1012 (18)	119
C18—H18 ⁱ ···O1	0.93	2.29	2.8870 (18)	122
C25—H25 ⁱ ···O3	0.93	2.50	2.9084 (19)	107
N1—H1N ⁱ ···O2	0.86	2.20	2.6146 (15)	109
C5—H5 ⁱ ···O4 ⁱ	0.93	2.58	3.4134 (19)	149
C30—H30B ⁱ ···O3 ⁱⁱ	0.96	2.42	3.254 (2)	145
C30—H30C ⁱ ···O1 ⁱⁱ	0.96	2.57	3.089 (2)	114

Symmetry codes: (i) $-x + 1, -y + 1, -z + 1$; (ii) $x, -y + \frac{3}{2}, z + \frac{1}{2}$.

Data collection: *RAPID-AUTO* (Rigaku 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1994); software used to prepare material for publication: *SHELXTL*.

The authors thank the Scientific Research Fund Projects of China West Normal University (No. 06B003) and the Youth Fund Projects of Sichuan Educational Department (No. 2006B039).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2130).

References

- Chen, Q.-Y. & Chen, C.-F. (2004). *Tetrahedron Lett.* **45**, 6493–6496.
Gunnlaugsson, T., Davis, A. P., O'Brien, J. E. & Glynn, M. (2005). *Org. Biomol. Chem.* pp. 48–56.
Gunnlaugsson, T., Lee, T. C. & Parkesh, R. (2003). *Org. Lett.* **5**, 4065–4068.
Rigaku (2004). *RAPID-AUTO* (Version 3.0). Rigaku/MSI Inc., The Woodlands, Texas, USA.
Rogers, M. E. & Averill, B. A. (1986). *J. Org. Chem.* **51**, 3308–3314.
Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
Siemens (1994). *SHELXTL*. Release 5.03. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

supplementary materials

Acta Cryst. (2007). E63, o3816 [doi:10.1107/S1600536807039311]

N,N'-Bis(2-methoxyphenyl)anthracene-1,8-dicarboxamide

E. Gao, R. He, H.-X. Zhang and D.-B. Qin

Comment

Anthracene as well as its derivatives and analogs have been widely used as signaling subunits for both cation (Gunnlaugsson *et al.*, 2003) and anion (Chen & Chen, 2004) sensing, due to their well known photophysical properties and high fluorescence. We report here the crystal structure of the title compound, (I).

In the molecule, the anthracene ring system are slightly non-planar with a maximum deviation of ± 0.036 Å. The anthracene group (C2...C15) and benzene group C17...C22 planes form a dihedral angle of $59.26(5)^\circ$, and the phenyl ring, C24...C29 makes a dihedral angle of $82.83(5)^\circ$ with the anthracene group. The dihedral angle between the two benzene rings is $42.62(5)^\circ$ (Fig. 1). In the molecule, there are four very weak intramolecular C—H...O hydrogen bonds. The crystal structure is stabilized by weak intermolecular C—H...O hydrogen bonds.

Experimental

The title compound was prepared according to the reported procedure of Rogers & Averill (1986) and Gunnlaugsson *et al.* (2005). Yellow single crystals suitable for X-ray diffraction were obtained by recrystallization from dimethyl sulfoxide.

Refinement

All H atoms were placed in calculated positions with C—H = 0.93 (aromatic CH) or 0.96 Å (methyl CH₃) and N—H = 0.86 Å, and refined using a riding model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier atom})$.

Figures

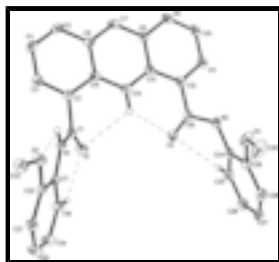


Fig. 1. The structure of (I), showing 30% probability displacement ellipsoids and the atomic numbering. H atoms involved in an intramolecular hydrogen bond are shown, while other have been omitted.

N,N'-Bis(2-methoxyphenyl)anthracene-1,8-dicarboxamide

Crystal data

C₃₀H₂₄N₂O₄

$M_r = 476.51$

$F_{000} = 1000$

$D_x = 1.363 \text{ Mg m}^{-3}$

supplementary materials

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 13.131$ (3) Å

$b = 11.864$ (2) Å

$c = 14.911$ (3) Å

$\beta = 91.13$ (3)°

$V = 2322.4$ (8) Å³

$Z = 4$

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 9633 reflections

$\theta = 3.1$ – 27.5 °

$\mu = 0.09$ mm⁻¹

$T = 153$ (2) K

Block, yellow

$0.30 \times 0.28 \times 0.28$ mm

Data collection

Rigaku R-Axis RAPID
diffractometer

Radiation source: Rotating Anode

Monochromator: graphite

$T = 153$ (2) K

ω scans

Absorption correction: none

18982 measured reflections

4318 independent reflections

3977 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.016$

$\theta_{\text{max}} = 25.5$ °

$\theta_{\text{min}} = 3.1$ °

$h = -15 \rightarrow 15$

$k = -14 \rightarrow 14$

$l = -17 \rightarrow 18$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.040$

$wR(F^2) = 0.107$

$S = 1.03$

4318 reflections

328 parameters

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0556P)^2 + 1.057P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.45$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Extinction correction: SHELXL97,
 $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0073 (9)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.80682 (7)	0.43852 (9)	0.32771 (6)	0.0313 (2)
O2	0.92280 (7)	0.28470 (9)	0.61312 (6)	0.0292 (2)
O3	0.69131 (8)	0.75454 (9)	0.28542 (7)	0.0346 (3)
O4	0.71944 (10)	0.92830 (10)	0.55977 (7)	0.0415 (3)
N1	0.83860 (8)	0.35283 (9)	0.46228 (7)	0.0225 (2)
H1N	0.8081	0.3267	0.5084	0.027*
N2	0.64594 (9)	0.86852 (10)	0.40038 (8)	0.0275 (3)

H2N	0.5991	0.8824	0.4383	0.033*
C1	0.77717 (10)	0.38556 (11)	0.39246 (8)	0.0218 (3)
C2	0.66880 (10)	0.34690 (11)	0.39944 (8)	0.0231 (3)
C3	0.65052 (12)	0.23509 (13)	0.41266 (10)	0.0332 (3)
H3	0.7051	0.1865	0.4229	0.040*
C4	0.55002 (12)	0.19140 (13)	0.41107 (11)	0.0385 (4)
H4	0.5394	0.1145	0.4185	0.046*
C5	0.46969 (11)	0.26096 (13)	0.39878 (10)	0.0321 (3)
H5	0.4040	0.2316	0.3983	0.038*
C6	0.48431 (10)	0.37878 (12)	0.38651 (9)	0.0275 (3)
C7	0.40171 (10)	0.45152 (13)	0.37357 (9)	0.0302 (3)
H7	0.3361	0.4220	0.3739	0.036*
C8	0.41418 (10)	0.56646 (13)	0.36024 (9)	0.0282 (3)
C9	0.32907 (10)	0.63914 (14)	0.34815 (10)	0.0327 (3)
H9	0.2635	0.6095	0.3491	0.039*
C10	0.34224 (11)	0.75113 (15)	0.33527 (11)	0.0374 (4)
H10	0.2859	0.7975	0.3262	0.045*
C11	0.44143 (11)	0.79795 (13)	0.33559 (10)	0.0340 (3)
H11	0.4495	0.8753	0.3284	0.041*
C12	0.52564 (10)	0.73085 (12)	0.34629 (9)	0.0260 (3)
C13	0.51545 (10)	0.61194 (12)	0.35810 (9)	0.0251 (3)
C14	0.59885 (10)	0.53955 (12)	0.36966 (8)	0.0245 (3)
H14	0.6644	0.5689	0.3671	0.029*
C15	0.58584 (10)	0.42404 (12)	0.38492 (9)	0.0244 (3)
C16	0.62893 (10)	0.78344 (11)	0.34033 (9)	0.0252 (3)
C17	0.94581 (9)	0.35625 (11)	0.46862 (8)	0.0212 (3)
C18	1.00793 (10)	0.39188 (12)	0.40017 (9)	0.0273 (3)
H18	0.9794	0.4213	0.3476	0.033*
C19	1.11299 (11)	0.38351 (15)	0.41045 (10)	0.0368 (4)
H19	1.1546	0.4066	0.3642	0.044*
C20	1.15621 (11)	0.34135 (15)	0.48846 (10)	0.0371 (4)
H20	1.2267	0.3353	0.4942	0.045*
C21	1.09525 (11)	0.30789 (13)	0.55847 (10)	0.0299 (3)
H21	1.1245	0.2806	0.6115	0.036*
C22	0.99064 (10)	0.31554 (11)	0.54865 (9)	0.0230 (3)
C23	0.96345 (12)	0.23011 (13)	0.69133 (9)	0.0331 (3)
H23A	1.0000	0.1637	0.6740	0.040*
H23B	0.9088	0.2094	0.7299	0.040*
H23C	1.0089	0.2806	0.7227	0.040*
C24	0.73476 (10)	0.93539 (11)	0.40486 (9)	0.0247 (3)
C25	0.78309 (12)	0.97149 (12)	0.32848 (10)	0.0313 (3)
H25	0.7574	0.9509	0.2723	0.038*
C26	0.86950 (12)	1.03800 (13)	0.33509 (11)	0.0365 (4)
H26	0.9021	1.0614	0.2835	0.044*
C27	0.90712 (12)	1.06945 (12)	0.41811 (12)	0.0357 (4)
H27	0.9649	1.1145	0.4224	0.043*
C28	0.85948 (12)	1.03441 (12)	0.49528 (11)	0.0339 (3)
H28	0.8853	1.0559	0.5512	0.041*
C29	0.77318 (11)	0.96719 (11)	0.48910 (9)	0.0283 (3)

supplementary materials

C30	0.76496 (18)	0.94020 (17)	0.64739 (11)	0.0562 (5)
H30A	0.8295	0.9023	0.6493	0.067*
H30B	0.7210	0.9076	0.6911	0.067*
H30C	0.7748	1.0187	0.6605	0.067*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0252 (5)	0.0426 (6)	0.0260 (5)	0.0029 (4)	-0.0006 (4)	0.0120 (4)
O2	0.0251 (5)	0.0380 (6)	0.0244 (5)	0.0034 (4)	-0.0023 (4)	0.0094 (4)
O3	0.0320 (6)	0.0323 (6)	0.0395 (6)	-0.0012 (4)	0.0045 (4)	-0.0114 (5)
O4	0.0618 (7)	0.0416 (6)	0.0211 (5)	-0.0214 (6)	-0.0003 (5)	-0.0019 (4)
N1	0.0200 (5)	0.0263 (6)	0.0211 (5)	0.0000 (4)	-0.0014 (4)	0.0042 (4)
N2	0.0291 (6)	0.0278 (6)	0.0256 (6)	-0.0028 (5)	0.0002 (5)	-0.0051 (5)
C1	0.0223 (6)	0.0217 (6)	0.0213 (6)	0.0036 (5)	-0.0018 (5)	-0.0004 (5)
C2	0.0236 (7)	0.0275 (7)	0.0179 (6)	-0.0005 (5)	-0.0041 (5)	0.0017 (5)
C3	0.0300 (8)	0.0293 (7)	0.0399 (8)	0.0003 (6)	-0.0076 (6)	0.0070 (6)
C4	0.0384 (9)	0.0278 (8)	0.0488 (9)	-0.0077 (6)	-0.0095 (7)	0.0106 (7)
C5	0.0263 (7)	0.0366 (8)	0.0330 (8)	-0.0109 (6)	-0.0056 (6)	0.0080 (6)
C6	0.0230 (7)	0.0334 (7)	0.0260 (7)	-0.0041 (6)	-0.0025 (5)	0.0013 (6)
C7	0.0214 (7)	0.0429 (8)	0.0262 (7)	-0.0058 (6)	-0.0021 (5)	0.0013 (6)
C8	0.0221 (7)	0.0366 (8)	0.0258 (7)	0.0008 (6)	-0.0023 (5)	-0.0052 (6)
C9	0.0189 (7)	0.0465 (9)	0.0324 (7)	0.0028 (6)	-0.0030 (5)	-0.0121 (7)
C10	0.0253 (7)	0.0446 (9)	0.0419 (8)	0.0114 (7)	-0.0076 (6)	-0.0073 (7)
C11	0.0323 (8)	0.0329 (8)	0.0366 (8)	0.0071 (6)	-0.0064 (6)	-0.0052 (6)
C12	0.0250 (7)	0.0298 (7)	0.0229 (6)	0.0028 (6)	-0.0044 (5)	-0.0076 (5)
C13	0.0218 (7)	0.0302 (7)	0.0232 (6)	0.0009 (5)	-0.0031 (5)	-0.0048 (5)
C14	0.0225 (6)	0.0308 (7)	0.0201 (6)	-0.0022 (5)	-0.0028 (5)	-0.0013 (5)
C15	0.0215 (6)	0.0294 (7)	0.0223 (6)	-0.0015 (5)	-0.0031 (5)	-0.0010 (5)
C16	0.0276 (7)	0.0229 (7)	0.0248 (7)	0.0033 (5)	-0.0055 (5)	-0.0019 (5)
C17	0.0203 (6)	0.0195 (6)	0.0237 (6)	0.0002 (5)	-0.0032 (5)	-0.0025 (5)
C18	0.0251 (7)	0.0331 (7)	0.0234 (6)	-0.0032 (6)	-0.0032 (5)	0.0009 (5)
C19	0.0237 (7)	0.0563 (10)	0.0306 (8)	-0.0059 (7)	0.0026 (6)	0.0024 (7)
C20	0.0181 (7)	0.0554 (10)	0.0378 (8)	0.0006 (7)	-0.0026 (6)	0.0000 (7)
C21	0.0256 (7)	0.0349 (8)	0.0289 (7)	0.0046 (6)	-0.0077 (5)	-0.0003 (6)
C22	0.0243 (6)	0.0212 (6)	0.0233 (6)	0.0014 (5)	-0.0012 (5)	-0.0011 (5)
C23	0.0354 (8)	0.0379 (8)	0.0257 (7)	0.0058 (6)	-0.0049 (6)	0.0092 (6)
C24	0.0289 (7)	0.0187 (6)	0.0264 (7)	0.0011 (5)	-0.0028 (5)	-0.0003 (5)
C25	0.0403 (8)	0.0271 (7)	0.0266 (7)	-0.0001 (6)	0.0000 (6)	0.0000 (6)
C26	0.0399 (9)	0.0279 (7)	0.0421 (9)	-0.0014 (6)	0.0104 (7)	0.0031 (6)
C27	0.0299 (8)	0.0223 (7)	0.0549 (10)	-0.0019 (6)	-0.0007 (7)	0.0007 (6)
C28	0.0393 (8)	0.0242 (7)	0.0378 (8)	-0.0020 (6)	-0.0126 (6)	-0.0020 (6)
C29	0.0380 (8)	0.0204 (6)	0.0263 (7)	-0.0016 (6)	-0.0037 (6)	0.0008 (5)
C30	0.0973 (16)	0.0474 (10)	0.0234 (8)	-0.0241 (10)	-0.0103 (9)	0.0007 (7)

Geometric parameters (\AA , $^\circ$)

O1—C1	1.2219 (16)	C11—H11	0.9300
O2—C22	1.3730 (17)	C12—C13	1.428 (2)

O2—C23	1.4283 (16)	C12—C16	1.4971 (19)
O3—C16	1.2187 (17)	C13—C14	1.3996 (19)
O4—C29	1.3603 (18)	C14—C15	1.400 (2)
O4—C30	1.433 (2)	C14—H14	0.9300
N1—C1	1.3610 (17)	C17—C18	1.3851 (19)
N1—C17	1.4096 (16)	C17—C22	1.4059 (18)
N1—H1N	0.8600	C18—C19	1.389 (2)
N2—C16	1.3647 (18)	C18—H18	0.9300
N2—C24	1.4110 (18)	C19—C20	1.378 (2)
N2—H2N	0.8600	C19—H19	0.9300
C1—C2	1.5007 (18)	C20—C21	1.386 (2)
C2—C3	1.363 (2)	C20—H20	0.9300
C2—C15	1.4359 (19)	C21—C22	1.381 (2)
C3—C4	1.418 (2)	C21—H21	0.9300
C3—H3	0.9300	C23—H23A	0.9600
C4—C5	1.349 (2)	C23—H23B	0.9600
C4—H4	0.9300	C23—H23C	0.9600
C5—C6	1.423 (2)	C24—C25	1.383 (2)
C5—H5	0.9300	C24—C29	1.3965 (19)
C6—C7	1.396 (2)	C25—C26	1.384 (2)
C6—C15	1.4379 (19)	C25—H25	0.9300
C7—C8	1.388 (2)	C26—C27	1.375 (2)
C7—H7	0.9300	C26—H26	0.9300
C8—C9	1.420 (2)	C27—C28	1.384 (2)
C8—C13	1.4361 (19)	C27—H27	0.9300
C9—C10	1.354 (2)	C28—C29	1.387 (2)
C9—H9	0.9300	C28—H28	0.9300
C10—C11	1.416 (2)	C30—H30A	0.9600
C10—H10	0.9300	C30—H30B	0.9600
C11—C12	1.369 (2)	C30—H30C	0.9600
C22—O2—C23	117.08 (11)	C2—C15—C6	117.41 (12)
C29—O4—C30	117.35 (13)	O3—C16—N2	123.05 (13)
C1—N1—C17	128.25 (11)	O3—C16—C12	122.97 (12)
C1—N1—H1N	115.9	N2—C16—C12	113.96 (12)
C17—N1—H1N	115.9	C18—C17—C22	119.18 (12)
C16—N2—C24	124.84 (12)	C18—C17—N1	124.14 (12)
C16—N2—H2N	117.6	C22—C17—N1	116.61 (12)
C24—N2—H2N	117.6	C17—C18—C19	119.67 (13)
O1—C1—N1	123.91 (12)	C17—C18—H18	120.2
O1—C1—C2	121.95 (11)	C19—C18—H18	120.2
N1—C1—C2	114.09 (11)	C20—C19—C18	120.72 (14)
C3—C2—C15	120.45 (13)	C20—C19—H19	119.6
C3—C2—C1	118.52 (12)	C18—C19—H19	119.6
C15—C2—C1	120.82 (12)	C19—C20—C21	120.35 (13)
C2—C3—C4	121.34 (14)	C19—C20—H20	119.8
C2—C3—H3	119.3	C21—C20—H20	119.8
C4—C3—H3	119.3	C22—C21—C20	119.29 (13)
C5—C4—C3	120.25 (14)	C22—C21—H21	120.4
C5—C4—H4	119.9	C20—C21—H21	120.4

supplementary materials

C3—C4—H4	119.9	O2—C22—C21	124.46 (12)
C4—C5—C6	120.74 (13)	O2—C22—C17	114.78 (11)
C4—C5—H5	119.6	C21—C22—C17	120.76 (13)
C6—C5—H5	119.6	O2—C23—H23A	109.5
C7—C6—C5	121.22 (13)	O2—C23—H23B	109.5
C7—C6—C15	119.00 (13)	H23A—C23—H23B	109.5
C5—C6—C15	119.76 (13)	O2—C23—H23C	109.5
C8—C7—C6	122.23 (13)	H23A—C23—H23C	109.5
C8—C7—H7	118.9	H23B—C23—H23C	109.5
C6—C7—H7	118.9	C25—C24—C29	119.59 (13)
C7—C8—C9	121.32 (13)	C25—C24—N2	121.84 (12)
C7—C8—C13	118.96 (13)	C29—C24—N2	118.56 (12)
C9—C8—C13	119.72 (14)	C24—C25—C26	120.45 (14)
C10—C9—C8	120.75 (14)	C24—C25—H25	119.8
C10—C9—H9	119.6	C26—C25—H25	119.8
C8—C9—H9	119.6	C27—C26—C25	119.89 (14)
C9—C10—C11	120.31 (14)	C27—C26—H26	120.1
C9—C10—H10	119.8	C25—C26—H26	120.1
C11—C10—H10	119.8	C26—C27—C28	120.43 (14)
C12—C11—C10	120.86 (15)	C26—C27—H27	119.8
C12—C11—H11	119.6	C28—C27—H27	119.8
C10—C11—H11	119.6	C27—C28—C29	119.95 (14)
C11—C12—C13	120.74 (13)	C27—C28—H28	120.0
C11—C12—C16	118.76 (13)	C29—C28—H28	120.0
C13—C12—C16	120.41 (12)	O4—C29—C28	125.39 (13)
C14—C13—C12	123.11 (13)	O4—C29—C24	114.92 (12)
C14—C13—C8	119.30 (13)	C28—C29—C24	119.68 (13)
C12—C13—C8	117.57 (12)	O4—C30—H30A	109.5
C13—C14—C15	121.52 (13)	O4—C30—H30B	109.5
C13—C14—H14	119.2	H30A—C30—H30B	109.5
C15—C14—H14	119.2	O4—C30—H30C	109.5
C14—C15—C2	123.62 (12)	H30A—C30—H30C	109.5
C14—C15—C6	118.97 (12)	H30B—C30—H30C	109.5
C17—N1—C1—O1	-10.7 (2)	C5—C6—C15—C14	177.79 (13)
C17—N1—C1—C2	166.99 (12)	C7—C6—C15—C2	179.34 (12)
O1—C1—C2—C3	124.44 (15)	C5—C6—C15—C2	-2.30 (19)
N1—C1—C2—C3	-53.27 (17)	C24—N2—C16—O3	2.2 (2)
O1—C1—C2—C15	-50.30 (18)	C24—N2—C16—C12	-176.26 (12)
N1—C1—C2—C15	131.99 (12)	C11—C12—C16—O3	-120.69 (16)
C15—C2—C3—C4	1.0 (2)	C13—C12—C16—O3	55.99 (19)
C1—C2—C3—C4	-173.74 (13)	C11—C12—C16—N2	57.75 (17)
C2—C3—C4—C5	-1.8 (2)	C13—C12—C16—N2	-125.58 (13)
C3—C4—C5—C6	0.5 (2)	C1—N1—C17—C18	-2.3 (2)
C4—C5—C6—C7	179.89 (14)	C1—N1—C17—C22	-179.21 (12)
C4—C5—C6—C15	1.6 (2)	C22—C17—C18—C19	2.1 (2)
C5—C6—C7—C8	-179.09 (14)	N1—C17—C18—C19	-174.81 (14)
C15—C6—C7—C8	-0.8 (2)	C17—C18—C19—C20	-0.8 (2)
C6—C7—C8—C9	-179.33 (13)	C18—C19—C20—C21	-0.8 (3)
C6—C7—C8—C13	1.1 (2)	C19—C20—C21—C22	1.0 (2)

C7—C8—C9—C10	179.94 (14)	C23—O2—C22—C21	-7.0 (2)
C13—C8—C9—C10	-0.5 (2)	C23—O2—C22—C17	172.60 (12)
C8—C9—C10—C11	-1.4 (2)	C20—C21—C22—O2	179.81 (14)
C9—C10—C11—C12	1.9 (2)	C20—C21—C22—C17	0.3 (2)
C10—C11—C12—C13	-0.5 (2)	C18—C17—C22—O2	178.59 (12)
C10—C11—C12—C16	176.17 (13)	N1—C17—C22—O2	-4.31 (17)
C11—C12—C13—C14	-179.75 (13)	C18—C17—C22—C21	-1.8 (2)
C16—C12—C13—C14	3.6 (2)	N1—C17—C22—C21	175.27 (12)
C11—C12—C13—C8	-1.3 (2)	C16—N2—C24—C25	39.1 (2)
C16—C12—C13—C8	-177.93 (11)	C16—N2—C24—C29	-142.06 (14)
C7—C8—C13—C14	-0.12 (19)	C29—C24—C25—C26	0.5 (2)
C9—C8—C13—C14	-179.70 (12)	N2—C24—C25—C26	179.32 (13)
C7—C8—C13—C12	-178.62 (12)	C24—C25—C26—C27	-0.7 (2)
C9—C8—C13—C12	1.81 (19)	C25—C26—C27—C28	0.4 (2)
C12—C13—C14—C15	177.20 (12)	C26—C27—C28—C29	0.0 (2)
C8—C13—C14—C15	-1.20 (19)	C30—O4—C29—C28	-11.8 (2)
C13—C14—C15—C2	-178.36 (12)	C30—O4—C29—C24	168.75 (15)
C13—C14—C15—C6	1.54 (19)	C27—C28—C29—O4	-179.47 (14)
C3—C2—C15—C14	-179.08 (13)	C27—C28—C29—C24	-0.1 (2)
C1—C2—C15—C14	-4.43 (19)	C25—C24—C29—O4	179.30 (13)
C3—C2—C15—C6	1.03 (19)	N2—C24—C29—O4	0.46 (19)
C1—C2—C15—C6	175.67 (12)	C25—C24—C29—C28	-0.1 (2)
C7—C6—C15—C14	-0.56 (19)	N2—C24—C29—C28	-178.98 (13)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C14—H14 \cdots O1	0.93	2.51	3.0581 (18)	118
C14—H14 \cdots O3	0.93	2.55	3.1012 (18)	119
C18—H18 \cdots O1	0.93	2.29	2.8870 (18)	122
C25—H25 \cdots O3	0.93	2.50	2.9084 (19)	107
N1—H1N \cdots O2	0.86	2.20	2.6146 (15)	109
C5—H5 \cdots O4 ⁱ	0.93	2.58	3.4134 (19)	149
C30—H30B \cdots O3 ⁱⁱ	0.96	2.42	3.254 (2)	145
C30—H30C \cdots O1 ⁱⁱ	0.96	2.57	3.089 (2)	114

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $x, -y+3/2, z+1/2$.

Fig. 1

